

March 31, 2011 – Highlights der Bioenergieforschung - Wieselburg

# Impact of Oxygenates in Diesel Fuel Blends on Engine Emissions and Combustion Properties



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& Automotive Technology



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# Impact of Oxygenates in Diesel Fuel Blends on Engine Emissions and Combustion Properties

## Outline

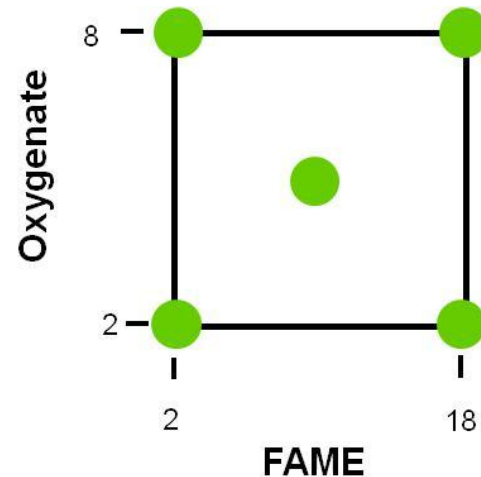
- ☐ Introduction and Task Description
- ☐ Technical Data of Test Engine
- ☐ Chemical Analysis
- ☐ Combustion Properties
- ☐ Consumption
- ☐ Engine Emissions
- ☐ Summary and Outlook

# Introduction

- Climate protection conventions (e.g. Kyoto Protocol)
  - ➡ Increased utilization of renewable primary products
  
- EU: Renewable Energy Directive (RED) and Fuels Quality Directive (FQD)
  - ➡ 10 energ.% biofuels from transport fuel pool
  - ➡ 20% green house gas savings
  
- Currently known substitutes: FAME, BTL or HVO
  - ➡ Bad quality (viscosity, oxidation stability) for higher blends (FAME)
  - ➡ Inadequate raw material base (BTL, HVO)
  - ➡ Expensive production process (BTL , HVO)
  - ➡ Maximum 7 vol.% FAME approved by car manufacturers

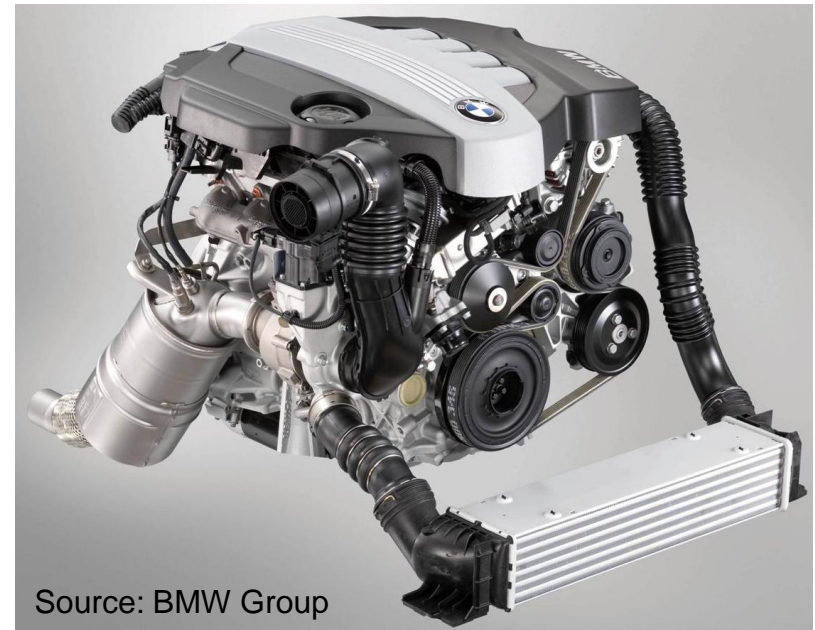
# Task Description

- Investigation of new biogenic oxygenates which fulfill ethical, ecological and economical requirements
  - ➔ Diesel substitute
  - ➔ Interaction with FAME containing diesel
- Investigated oxygenates
  - ➔ Glyme, Alcohol, Polyether, Tributylcitrat, Levulinat, Valeriat
- Design of Experiments of fuel blends
- Presented oxygenates
  - ➔ Glyme (Tetra-Glyme)
  - ➔ Alcohol (Butanol)



# Technical Data of Test Engine

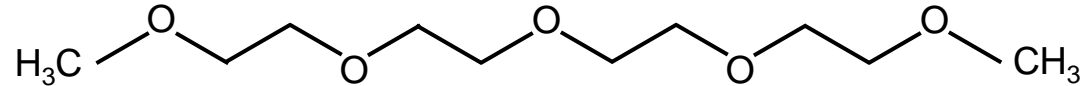
Diesel engine N47 D20 U1 (BMW)	
Cylinder	4 inline
Bore [mm]	84
Stroke [mm]	90
Displacement [cm <sup>3</sup> ]	1995
Compression Ratio	16
Power [kW]	105 at 4000 [rpm]
Max. Torque [Nm]	320 at 1750 [rpm]
Injection System	Common Rail
Turbo Charging	Var. Turbine Geometry



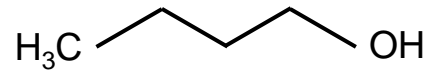
Source: BMW Group

# Chemical Analysis

Tetra-Glyme (C<sub>10</sub>H<sub>22</sub>O<sub>5</sub>)



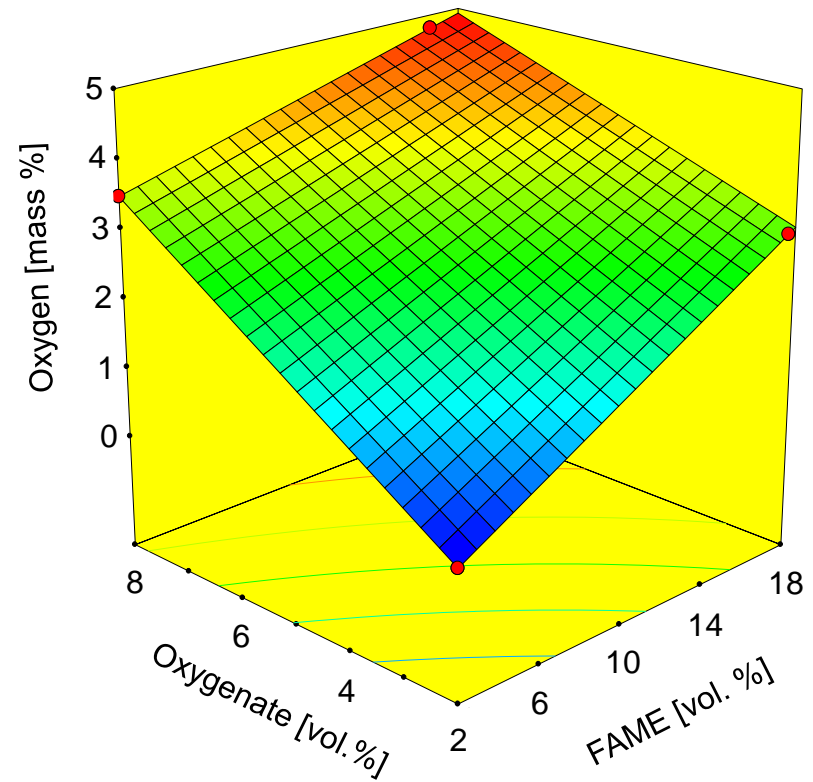
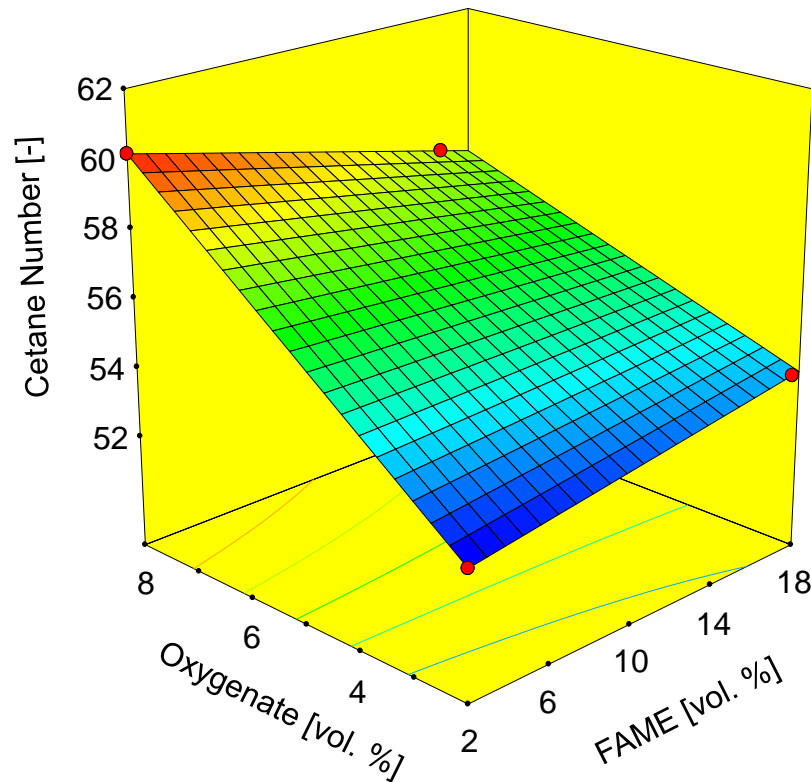
Butanol (C<sub>4</sub>H<sub>10</sub>O)



		Oxigenat	FAME	B0	C	H	O	C/H	CN	Density	Flash Point	Heat Value
		[vol. %]			[mass %]			[-]	[-]	[kg/m <sup>3</sup> ]	[°C]	[MJ/kg]
<b>B0</b>	-	0	0	100	86	14	0	6,14	52,1	829	61	43,5
<b>B18</b>	-	0	18	82	84,5	13,5	1,8	6,26	54,1	838,6	64,5	41,955
<b>Glyme</b>	Tetra-Glyme	8	2	90	82,9	13,4	3,5	6,19	60,2	842,9	63,5	41,209
		8	16,56	75,44	81,6	13,1	4,8	6,23	57,8	851	66,5	40,431
<b>Alcohol</b>	Butanol	8	2	90	84,7	13,7	1,6	6,18	48,1	828,2	39,5	42,201
		8	16,56	75,44	82,7	13,5	3,2	6,13	49,9	835,8	38,5	41,083

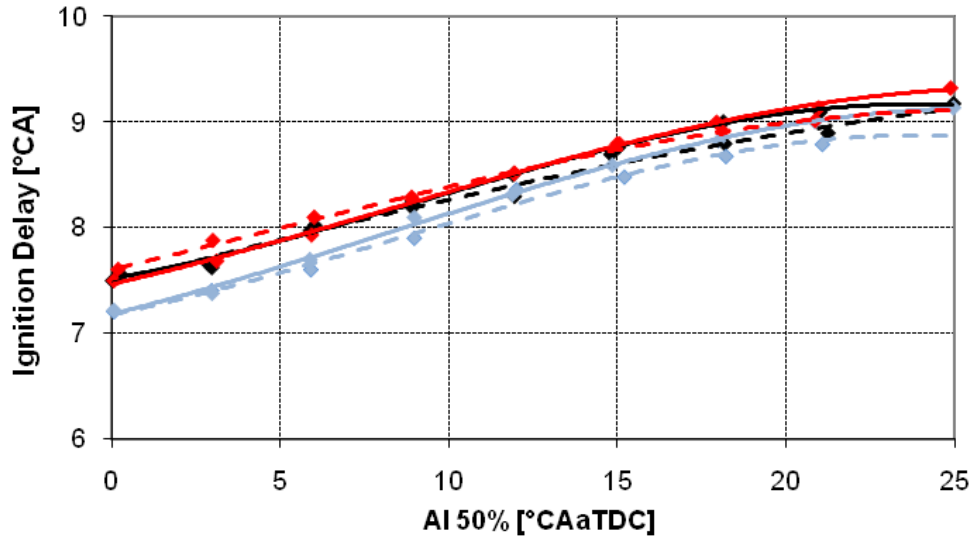
# Chemical Analysis

## Tetra-Glyme

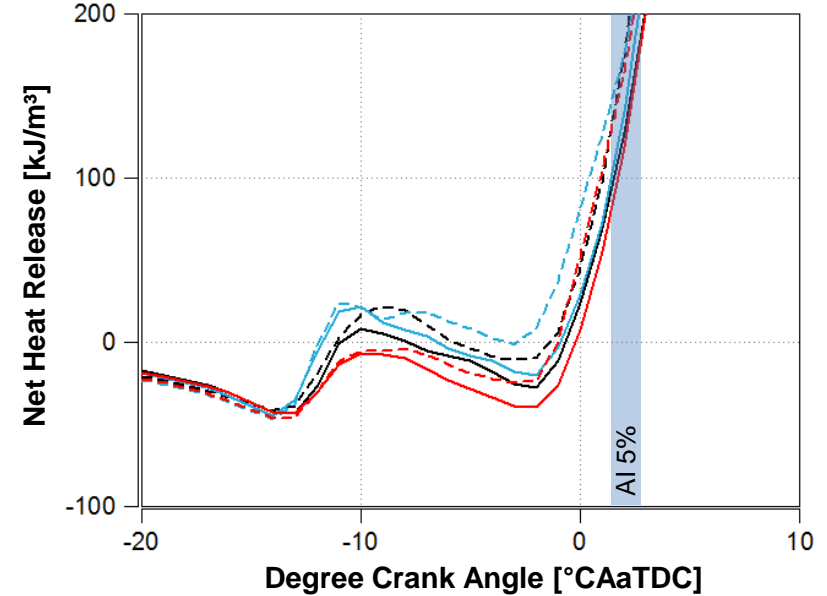


# Combustion Properties

Variation of AI 50% at 2000 rpm and 15 bar BMEP



Loadpoint: 2000 rpm, 15 bar BMEP, 12°CAaTDC



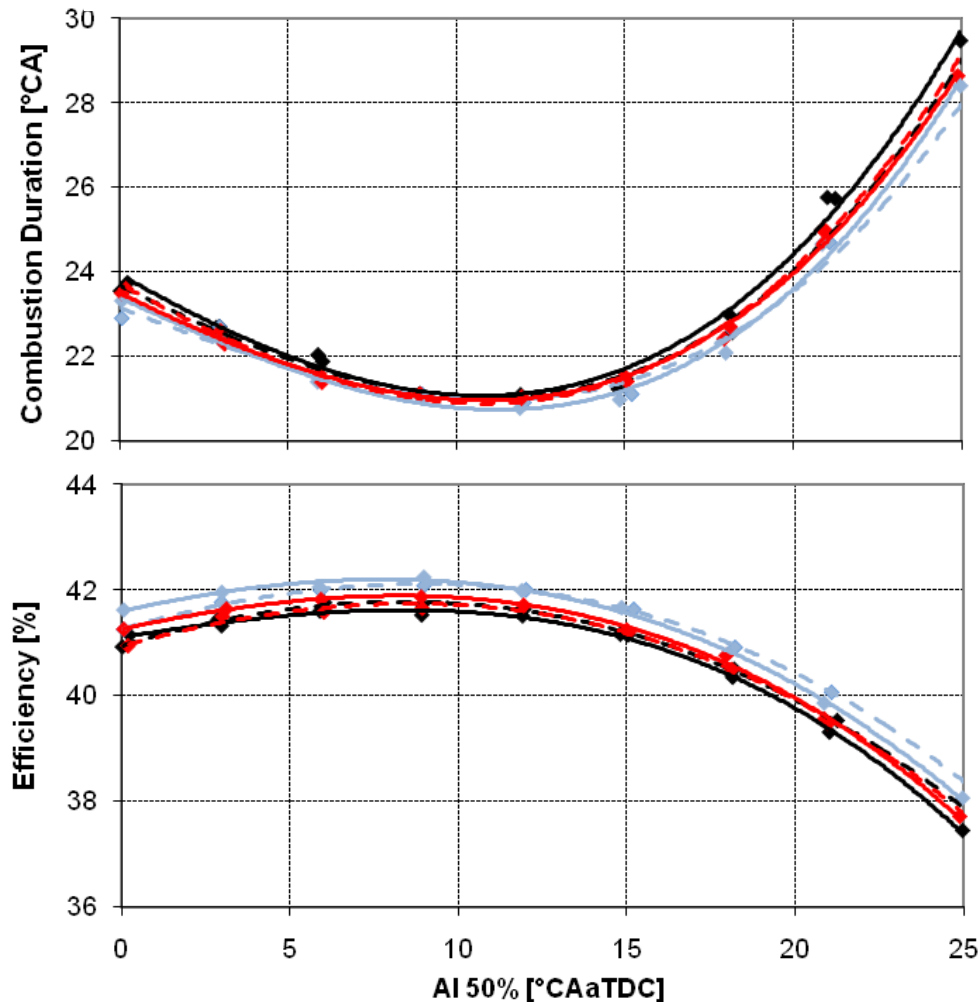
- Additivation of Tetra-Glyme increases CN
  - ➡ higher and faster net heat release
- Additivation of Butanol decreases CN
  - ➡ lower and slower net heat release
- Additivation of FAME has only minor impact on CN and combustion properties

		CN
		[-]
—	B0	52,1
- - -	B18	54,1
—	Tetra-Glyme	60,2
- - -	Tetra-Glyme + FAME	57,8
—	Butanol	48,1
- - -	Butanol + FAME	49,9

# Combustion Properties

		CN	O
		[-]	[mass %]
—	B0	52,1	0
- - -	B18	54,1	1,8
—	Tetra-Glyme	60,2	3,5
- - -	Tetra-Glyme + FAME	57,8	4,8
—	Butanol	48,1	1,6
- - -	Butanol + FAME	49,9	3,2

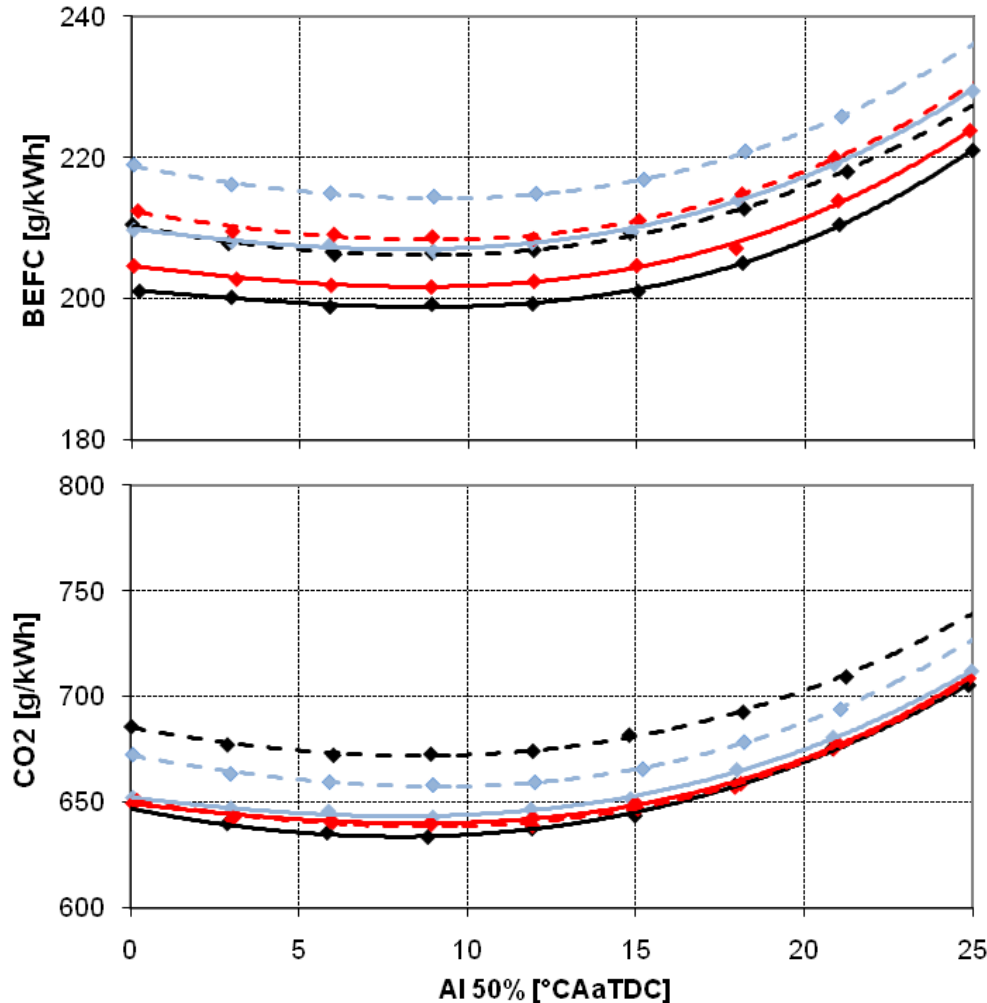
Variation of AI 50% at 2000 rpm and 15 bar BMEP



- Additivation of Tetra-Glyme reduces combustion duration  
➡ Increase of efficiency
- Additivation of butanol has only minor impact on combustion duration and efficiency
- Additivation of FAME has only minor impact on combustion duration and efficiency

# Consumption

## Variation of AI 50% at 2000 rpm and 15 bar BMEP



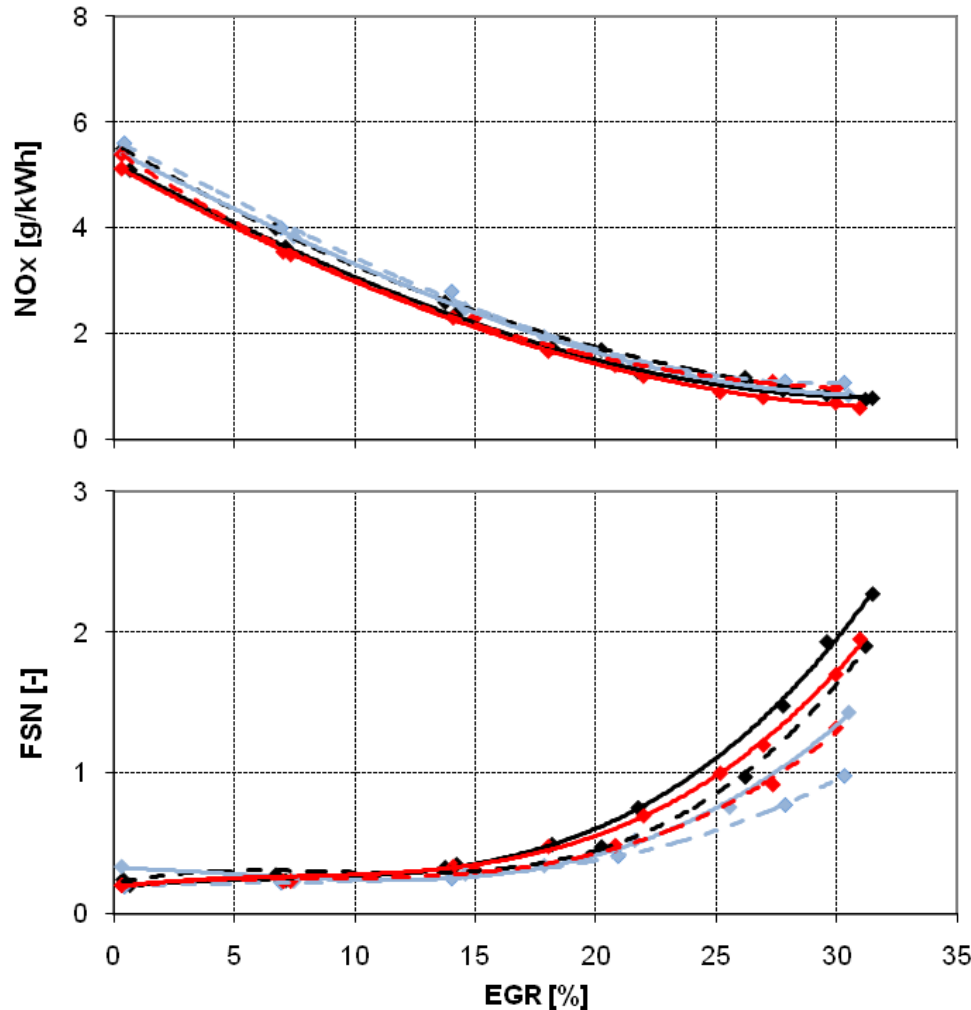
		Density	Heat Value	C
		[kg/m <sup>3</sup> ]	[MJ/kg]	[mass %]
—	B0	829	43,5	86
- - -	B18	838,6	41,955	84,5
—	Tetra-Glyme	842,9	41,209	82,9
- - -	Tetra-Glyme + FAME	851	40,431	81,6
—	Butanol	828,2	42,201	84,7
- - -	Butanol + FAME	835,8	41,083	82,7

- Additivation of oxygenate decreases heat value
  - ➡ Increase of gravimetric consumption
- Additivation of FAME further decreases heat value
  - ➡ Further increase of gravimetric consumption
- CO<sub>2</sub>-emissions dependent on relation between gravimetric consumption and mass of carbon in fuel blend

# Limited Emissions (NO<sub>x</sub> and Particle)

		CN	O
		[-]	[mass %]
—	B0	52,1	0
- - -	B18	54,1	1,8
—	Tetra-Glyme	60,2	3,5
- - -	Tetra-Glyme + FAME	57,8	4,8
—	Butanol	48,1	1,6
- - -	Butanol + FAME	49,9	3,2

Variation of EGR at 2000 rpm and 1 bar BMEP

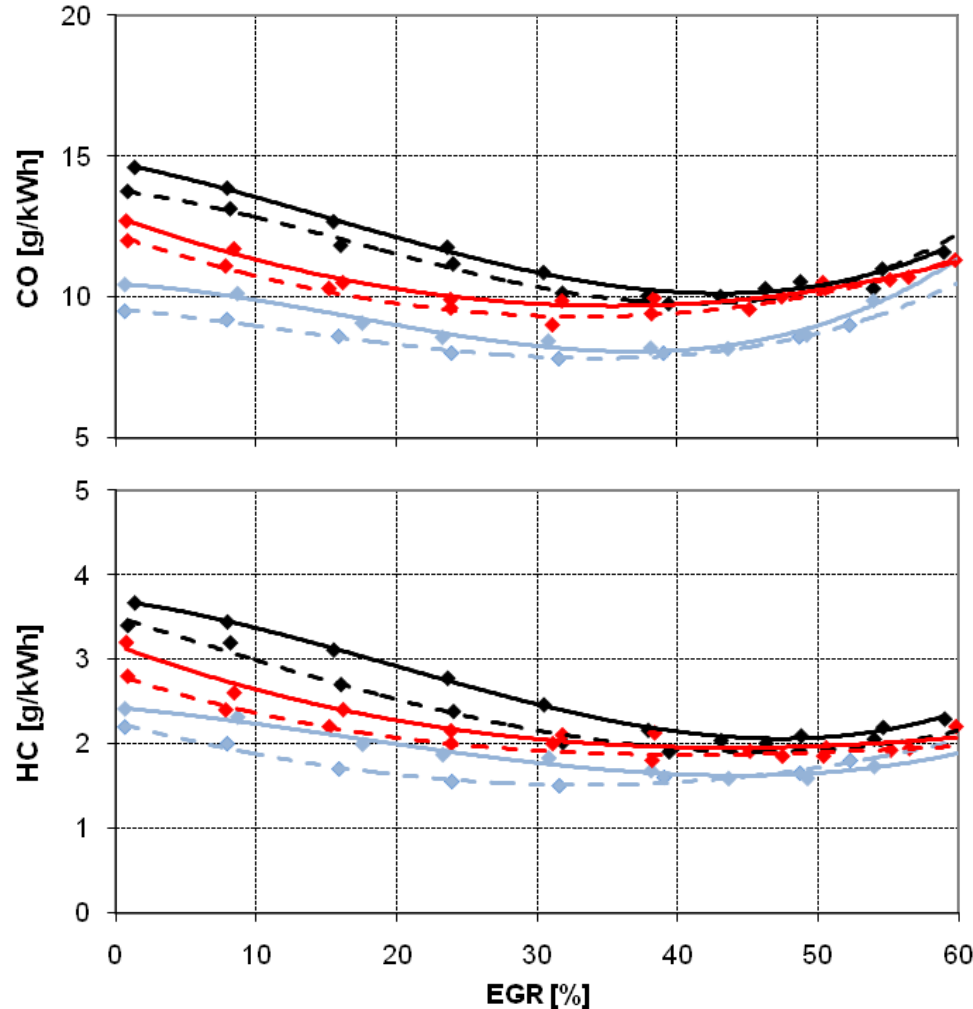


- Additivation of Tetra-Glyme decreases combustion duration
  - ➡ Higher combustion chamber temperatures
  - ➡ Higher NO<sub>x</sub>-emissions
- Additivation of oxygenate and FAME increases oxygen content
  - ➡ Reduction of substoichiometric zones
  - ➡ Decrease of particle emissions

# Limited Emissions (CO and HC)

		O
		[mass %]
—	B0	0
- - -	B18	1,8
—	Tetra-Glyme	3,5
- - -	Tetra-Glyme + FAME	4,8
—	Butanol	1,6
- - -	Butanol + FAME	3,2

Variation of EGR at 1500 rpm and 6,5 bar BMEP



- Additivation of oxygenate and FAME increases oxygen content
  - ➡ More homogeneous mixture preparation
  - ➡ Decrease of CO- and HC-emissions

# Summary and Outlook

- The additivation of the investigated oxygenates to diesel fuel leads to:
  - ➡ a slight decrease of the combustion duration (Tetra-Glyme)
  - ➡ an increase of the gravimetric fuel consumption
  - ➡ a high decrease of the CO-, HC- and particle emissions
  
- The additivation of FAME to the investigated diesel-oxygenate blends:
  - ➡ has only minor impact on the combustion properties
  - ➡ increases further the gravimetric fuel consumption
  - ➡ leads to a further decrease of the CO-, HC- and particle emissions
  
- Biogenic oxygenates deliver an ecological alternative as a diesel substitute and enable a high decrease of the CO-, HC- and particle emissions in combination with FAME. In this context further investigations (Tributylcitrat, Valeriat, Levulinat) will be undertaken.

# Thank you for your attention!



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